

Low Energy Nuclear Reaction Investigations at the Naval Research Laboratory

Off the chart!

D.A. Kidwell, D.D. Dominguez, K.S. Grabowski, and D.L. Knies U.S. Naval Research Laboratory

Washington, DC 20375

A.E. Moser, C. Cetina, and J.H. He

NOVA Research, Inc. Alexandria, VA

G.K. Hubler and C.A. Carosella

U.S. Naval Research Laboratory (Retired) Washington, DC 20375



Applying the Scientific Method to Understanding Anomalous Heat Effects: Opportunities and Challenges

18th International Conference on Condensed Matter Nuclear Science – Columbia, Missouri July 22-26, 2013

The Prime Directive



Simplified steps in scientific method:

- 1. Make some observation
- Make-up some hypothesis refuting
 Collect a lot of data supporting that hypothesis
- 4. Make a few presentations and write a few papers
- 5. Win a Nobel prize or make lots of money

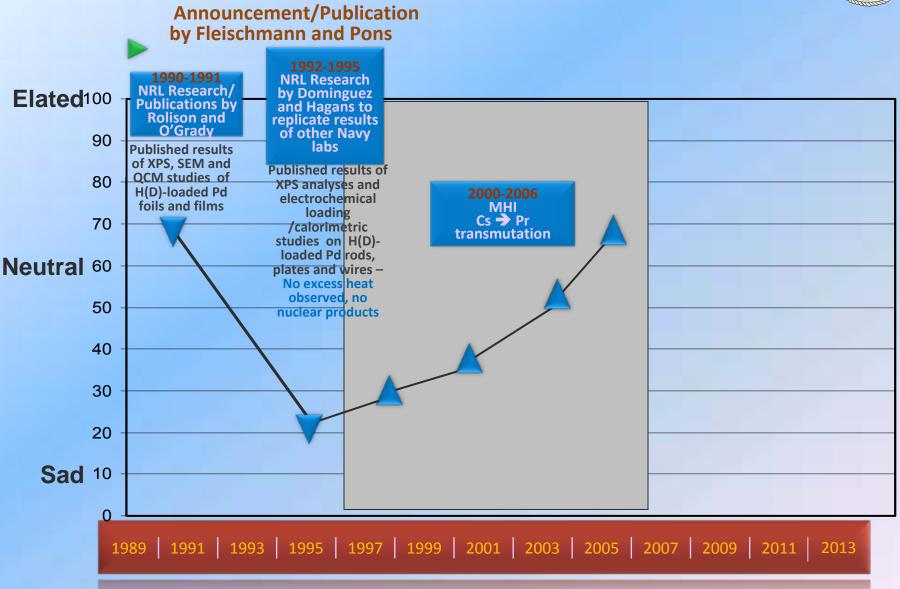
In science, the Prime Directive is to prove yourself wrong!

Logically, something only can be falsified

"There is no surer way to screw-up an experiment than to be certain of its outcome" Stuart Firestein – Ignorance: How it Drives Science

What Has NRL Observed?

Chart of Happiness





Prima-Facie Evidence that LENR Occurs

Something non-conventional is happening



- 1. Radiation on demand
- 2. Transmutation
 - Production of a radioactive isotope
 - Isotopic pattern change
 - Production of element not originally present
- 3. Definitive demonstration of excess energy
 - Self-sustained energy production beyond known chemistry

Radiation on Demand

Short answer is no definitive signa



Advantages:

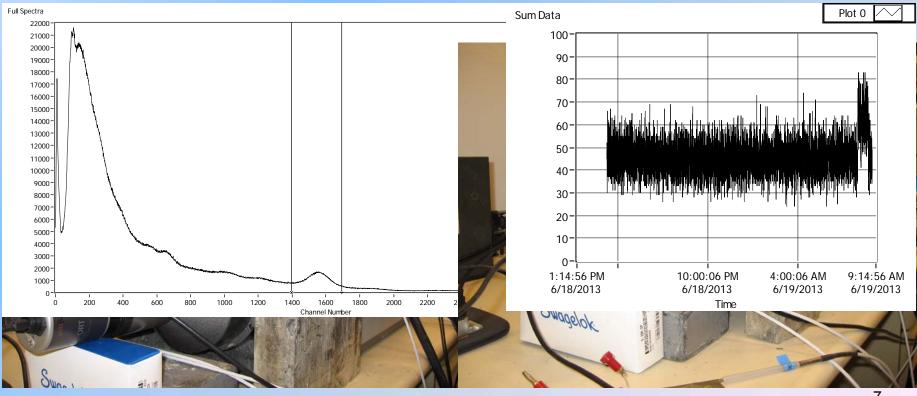
- The most sensitive over a broad range of energies
- Rugged and non-cooled
- Provide some spectral information but not highly resolved



Nal Scintillation Detectors

Advantages:

- The most sensitive over a broad range of energies
- Rugged and non-cooled
- Provide some spectral information but not highly resolved











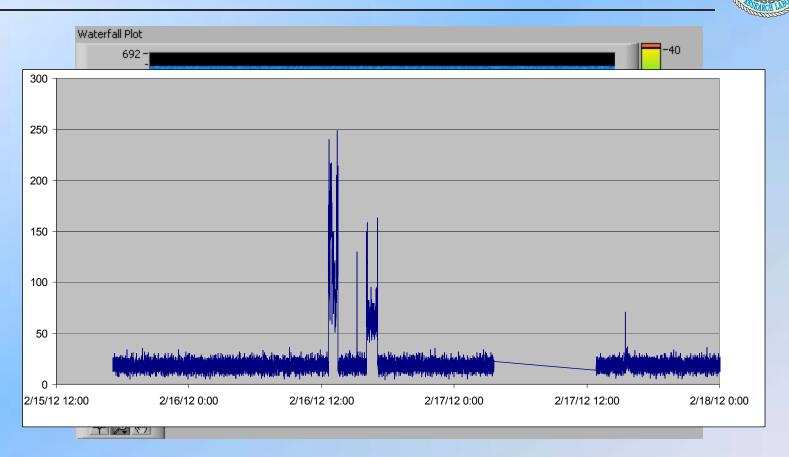
Yan Kurochov, "Method and Apparatus for Energy Generation", WO 98/15986 ➡Theory requires large thermal gradient

Rossi-Like Device



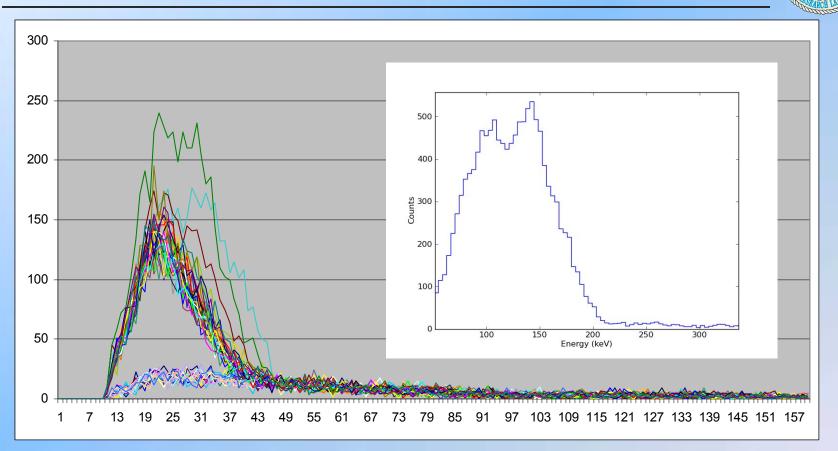


Radiation as Measured by a Nal Detector Les Case material



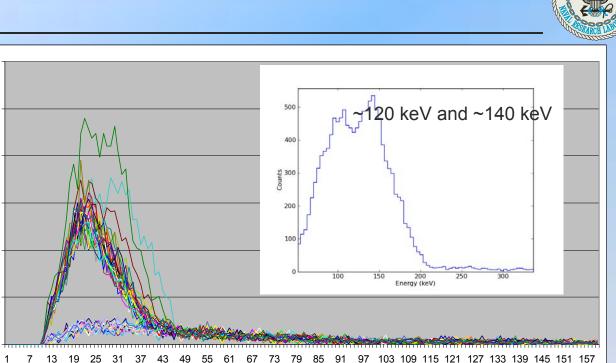
- Clear indication of three radiation events
 - 12:30, 16:07, 15:38 next day

Radiation as Measured by a Nal Detector Spectra



- Spectra substantially above background
 - Looks like two peaks at ~120 keV and ~140 keV

Radiation as Measured by a Nal Detector Explanation



Someone using a source nearby? – no

300

250

200

150

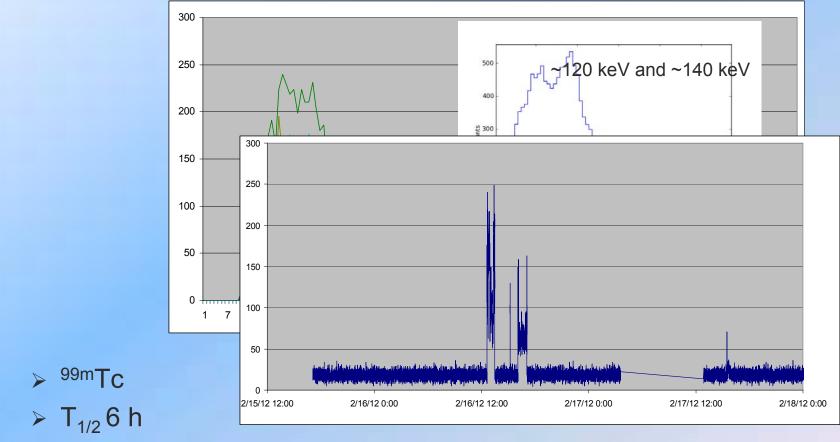
100

50

Something not right:

- System running for several days before radiation appeared
 - Nothing being done to change conditions
- Radiation never seen before
 - Conditions not correct?

Radiation as Measured by a Nal Detector Explanation



Gamma at 140.3 keV





- > Evidence for radiation weak
- For NRL, Chart of Happiness for radiation on demand is:

Off the chart!

Transmutation Isotopic Anomalies

Be Wary of Cluster Ions Secondary Ion Mass Spectrometry (SIMS) Inductively Coupled Plasma Mass Spectrometry (ICP-MS)

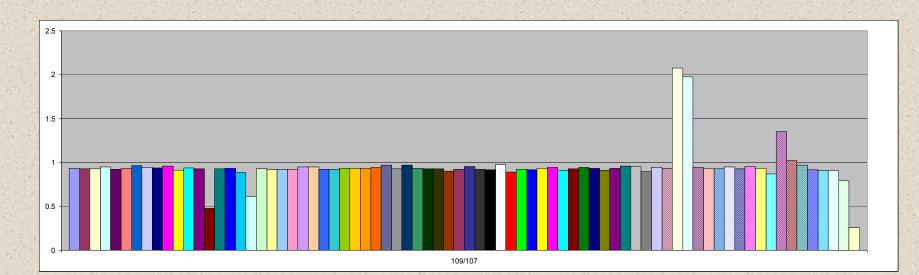
Short answer is **no definitive result**

MORPH[™] Ordering Information

Speculation

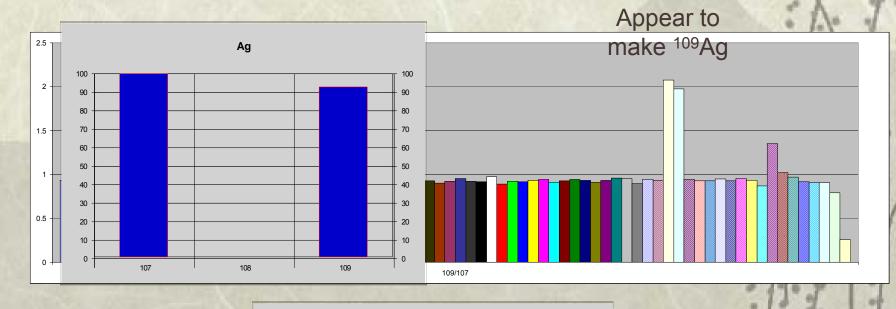
105 Pd +D \rightarrow 107 Ag

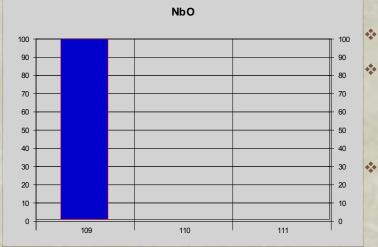
- Silver alleged to be created in some heat producing reactions
 - Two isotopes ¹⁰⁷Ag and ¹⁰⁹Ag
- BUT Silver is a trace contaminant in palladium
- How does this ratio vary across samples by ICP-MS?
 - How does ICP-MS Work?



Speculation

 $^{105}Pd + D \rightarrow ^{107}Ag$





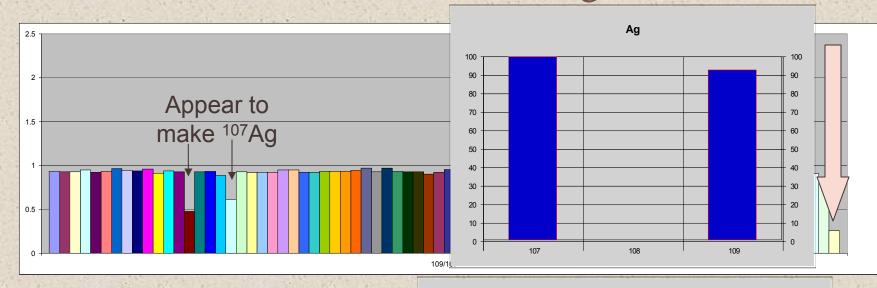
Niobium present
 m/z 109 Likely NbO

interferent

Good Thing because nuclear chemistry unclear

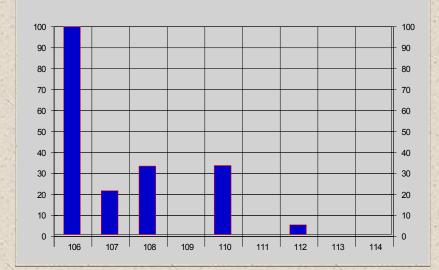
Speculation

$^{105}Pd + D \rightarrow ^{107}Ag$



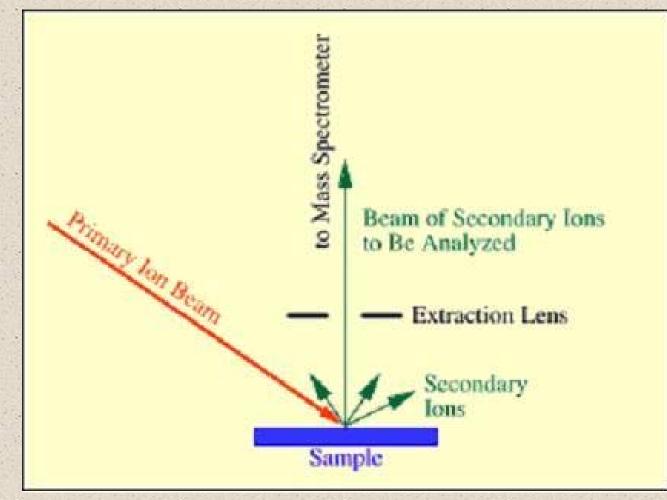
Zirconium present

* m/z 107 Likely ZrO interferent



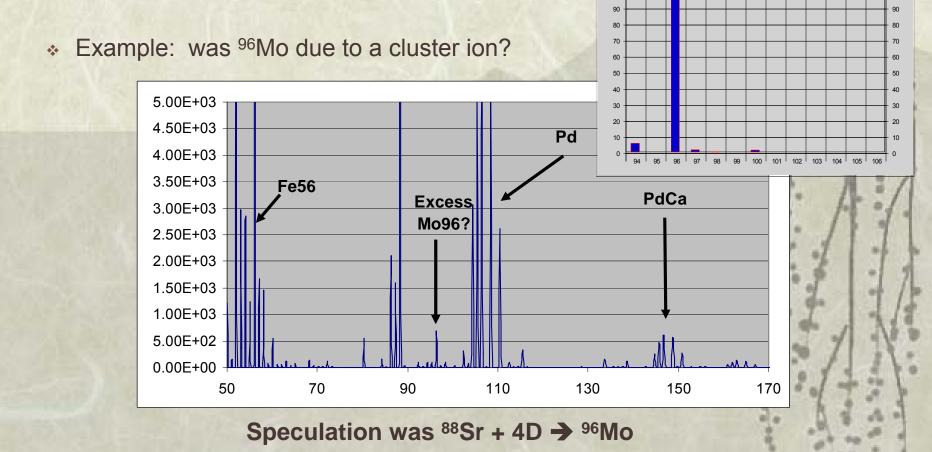
ZrO

Be Wary of Cluster Ions Secondary Ion Mass Spectrometry (SIMS)



From: http://serc.carleton.edu/details/images/8453.html

Be Wary of Cluster Ions Secondary Ion Mass Spectrometry (SIMS)

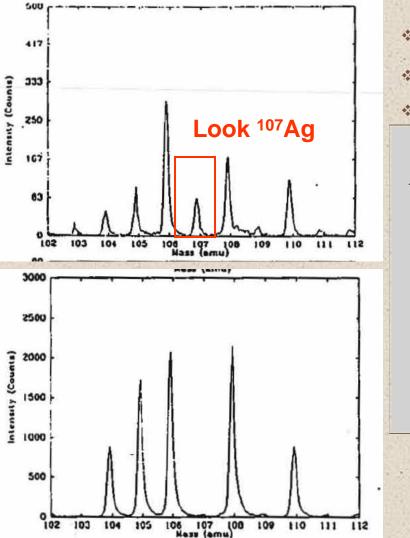


100

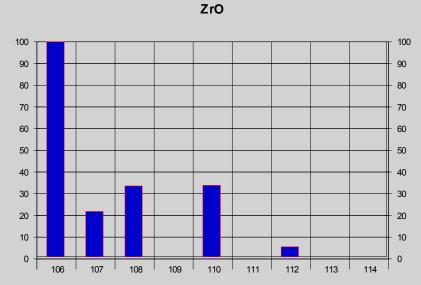
Solution State State

Be Wary of Cluster Ions Secondary Ion Mass Spectrometry (SIMS)

Data from: D.R. Rolison and W.E. O'Grady, "Mass/charge anomalies in Pd after electrochemical loading with deuterium", EPRI-NSF Workshop on Anomalous Effects in Deuterated Metals, Oct 16-18, 1989.



- ✤ Observe large ¹⁰⁶Pd enhancement
- Zirconium present
- * m/z 106 Likely ZrO interferent



Conclusions for Transmutation based on isotopic patterns

- Molecular interferents will confuse isotopic patterns
 - -SIMS be very wary
 - -ICP-MS better but not perfect
- For NRL, Chart of Happiness for Transmutation based on isotopic patterns is:

Off the chart!

Transmutation

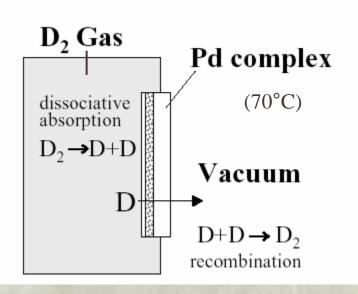
Evaluation of the Claim of Transmutation of Cesium to Praseodymium with Mitsubishi Heavy Industries (MHI)



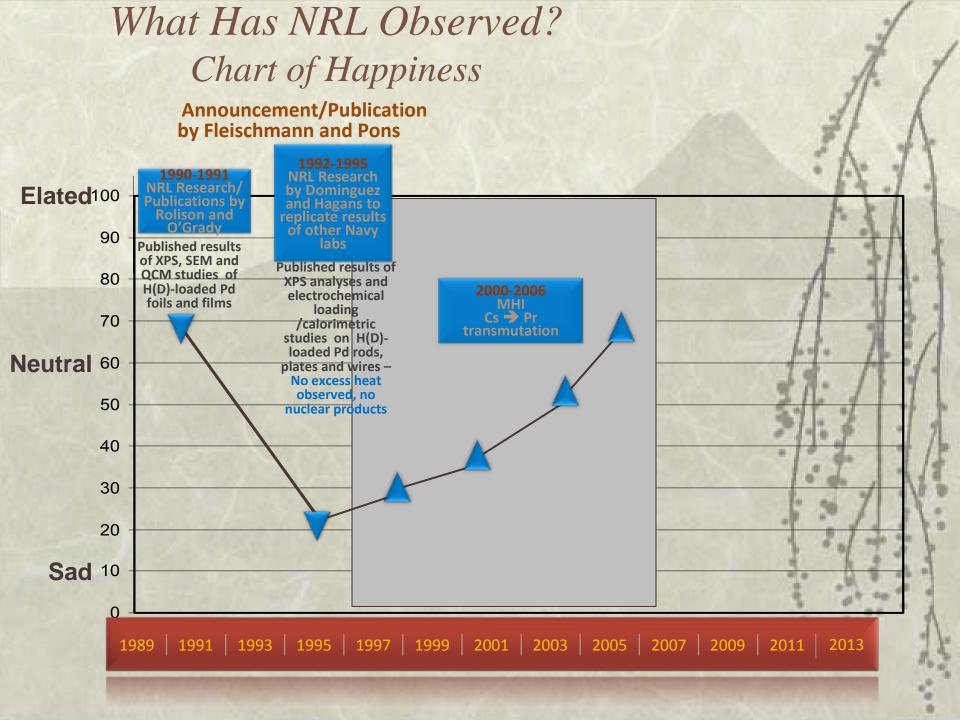
Summary of MHI Claims

- By permeating Deuterium though a Pd complex foil, various elemental transmutations can be made to happen
 - -Reported transmutations:

• ⁸⁸ Sr → ⁹⁶ Mo	Addition of 4D
• ¹³³ Cs → ¹⁴¹ Pr	Addition of 4D
• ¹³⁷ Ba → ¹⁴⁹ Sr	n Addition of 6D



Y. Iwamura , M. Sakano and T. Itoh, "Elemental Analysis of Pd Complexes: Effects of D₂ Gas Permeation", *Jpn. J. Appl. Phys.*, **41** (2002) 4642–4650.



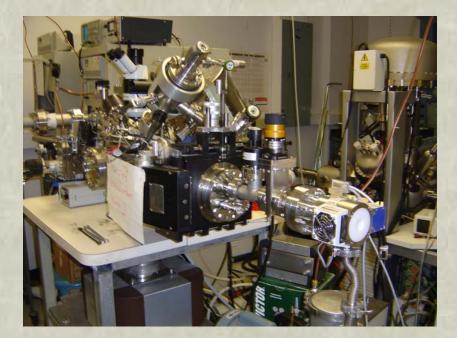
NRL Research Effort

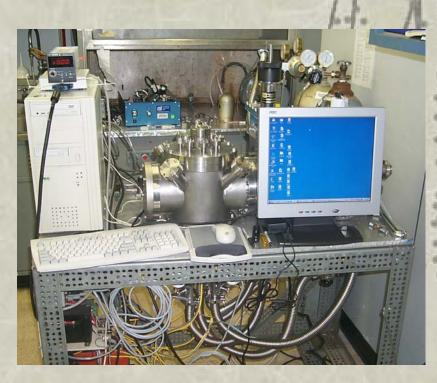
- Collaborated with MHI to verify Pr present on their permeated samples
 - Initially with Accelerator Mass Spectrometry
 - Eventually with ICP-MS at part per quadrillion levels (10-15)
 - NRL convinced Pr was present on some samples

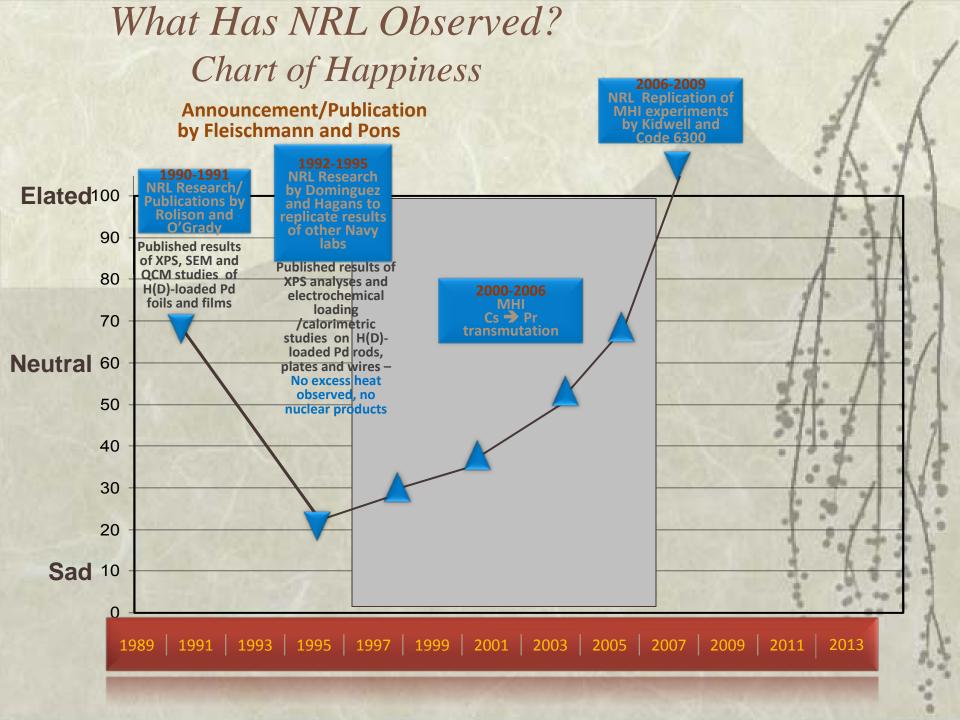
Modified XPS and 5-Sample Chambers @ NRL

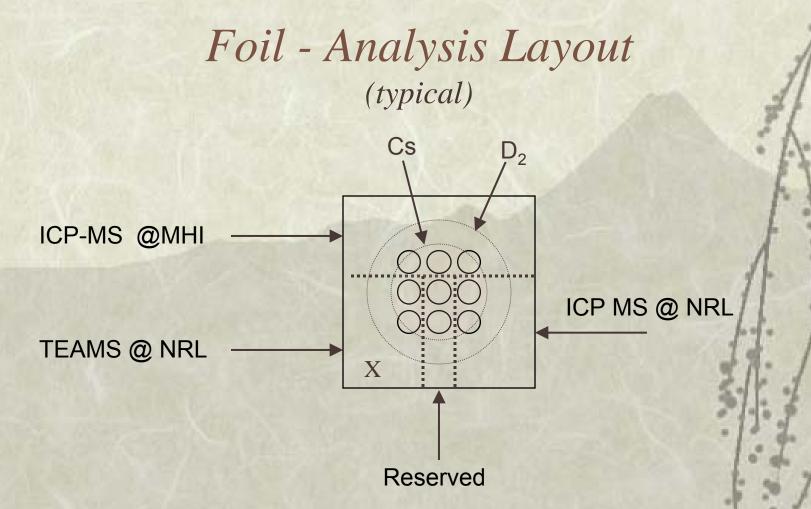
NRL attempted to independently reproduce result Unsuccessful at producing Pr

Therefore, performed joint research with MHI





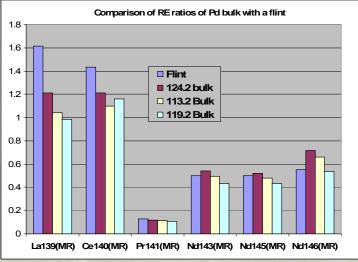




Note: Lower 2/3 Sent to NRL for Analysis

Pr is mono isotopic so what would be the signature?

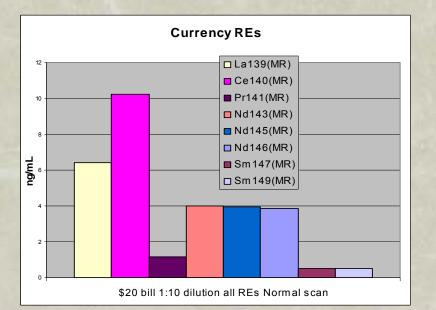
Rare Earths in the Environment Rare earths are not so rare

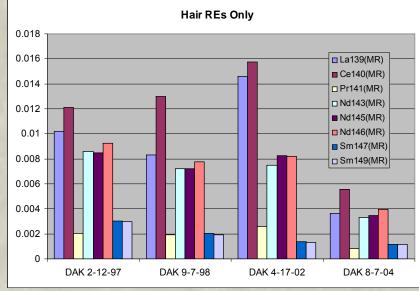


Environmental sources of Pr

- Flints "Misch Metal"
- Specialty glazes on ceramics

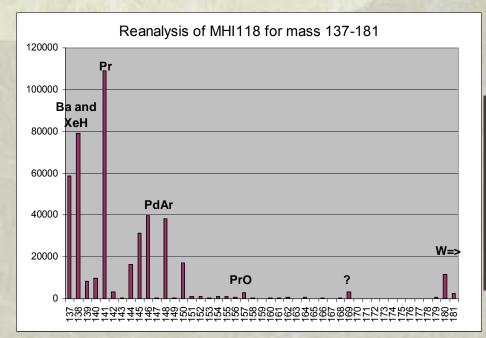
- Phosphors in CRTs (tends to be with other REs)



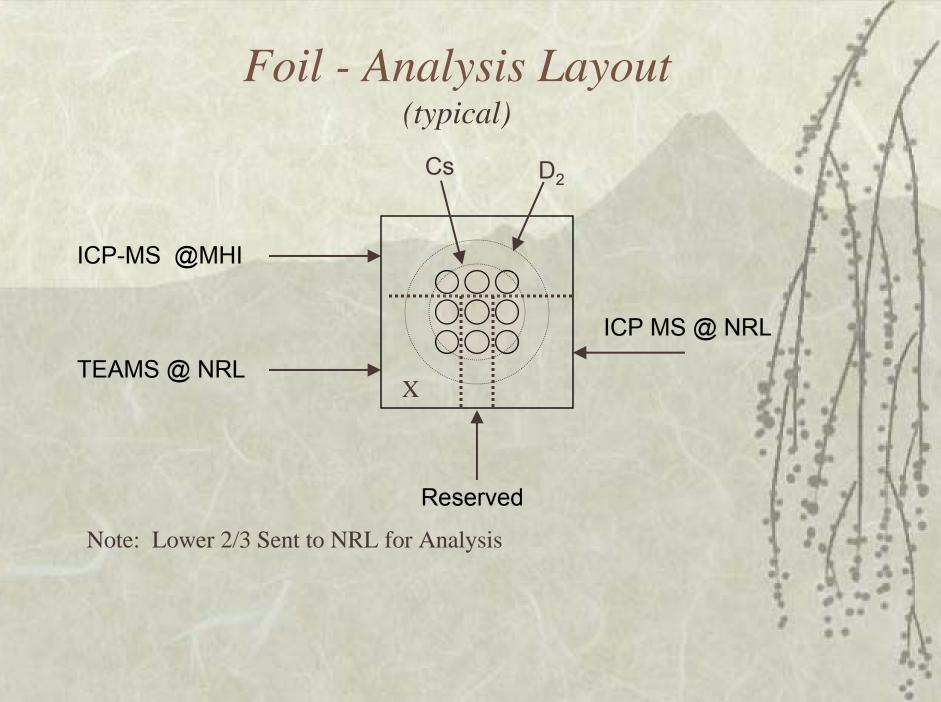


Results - Molecular Interferences or Misch Metal Contamination?

- 15 extracts & 6 blanks were sent from MHI
- 18 Confirmed negative for Pr LOD <0.045 ng/cm
 Includes 6 controls with the three positives (three sets of three)



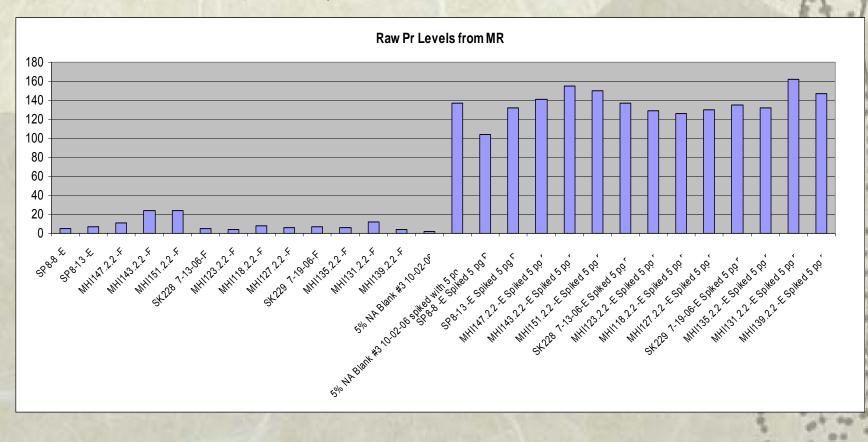
Sample	Toray ng/cm ²	NRL ng/cm ²	% Differen ce
MHI118	20	19	5.00%
MHI131	9.1	9.2	-1.50%
MHI143	25	24	5.54%



Big Issue – Could not find Pr at NRL on Our part of the Foils!

Example Showing That NRL Could Have Observed Pr, If Present

One series spiked with equivalent of 170 pg/cm² Pr (~100x lower than expected)



Possible Explanations for failure to find Pr

- Pr not there we clearly have the required sensitivity and specificity
 - Mistakes at MHI (blanks mitigate this)
- Wrong part of foil
 - Systematic issues statistically improbable
- Pr migrates into the interior <u>bulk analysis should find it!</u>
 Have done bulk analysis on MHI118 and essentially negative
- Pd catalyzes reverse nuclear reaction with time?
 - -Real stretch of science
- Pr lost in transit or handling
 - Film unstable Pr should be in boxes checked and no
- Not etching enough Pd foil
 - Loss of deuterium with time causes lower Pd etching
 - Travel to MHI to extract a freshly prepared sample

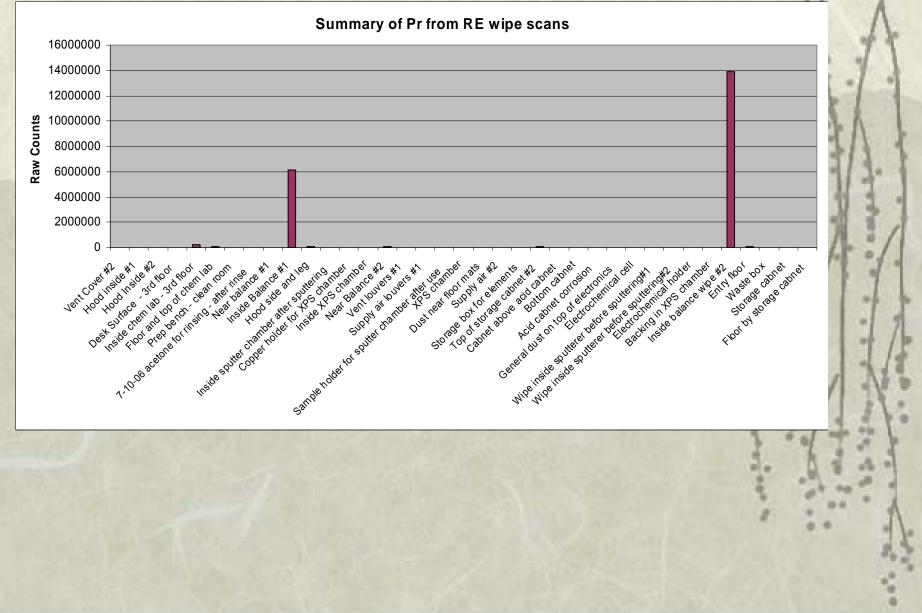
Months worth of work (frustration) later.,

- Traveled to MHI to observe process and extract a freshly prepared sample
- Participated in extraction using their equipment on a fresh sample
- Take split sample extracts

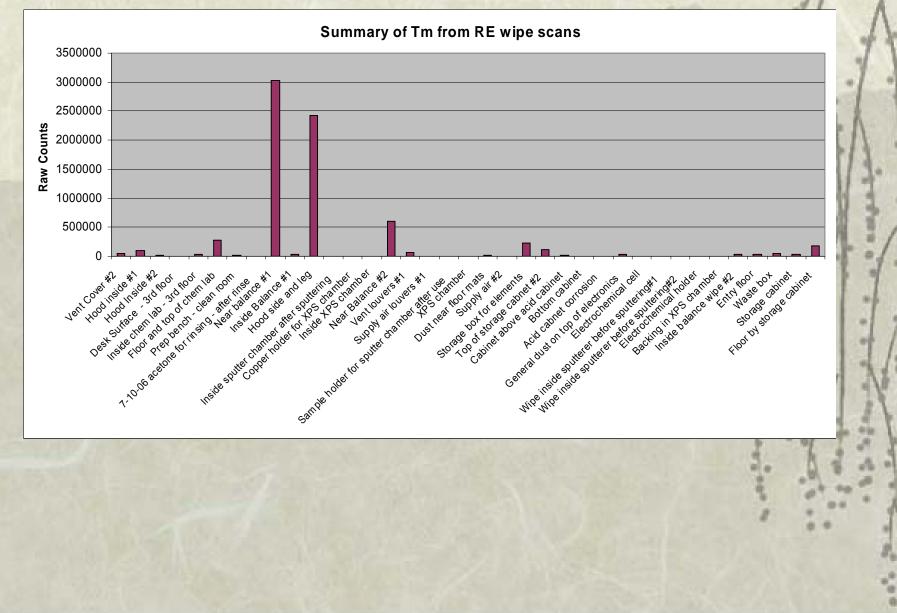
 Samples extracted at MHI in the presence of NRL personnel were <u>negative</u> at NRL and with a commercial laboratory

On whim, do environmental survey

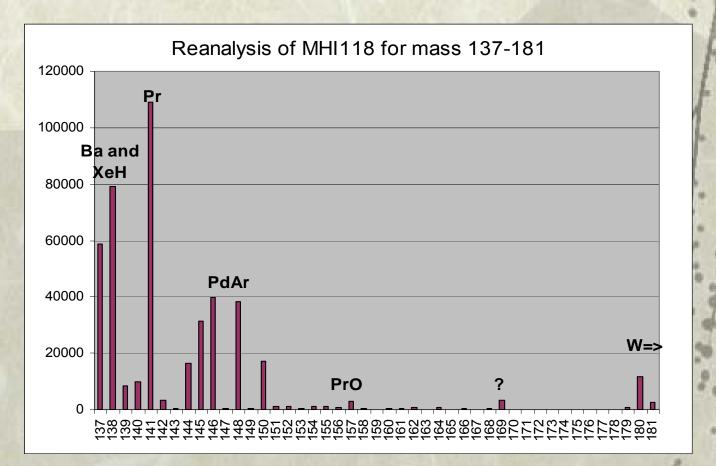
Results from Environmental Survey



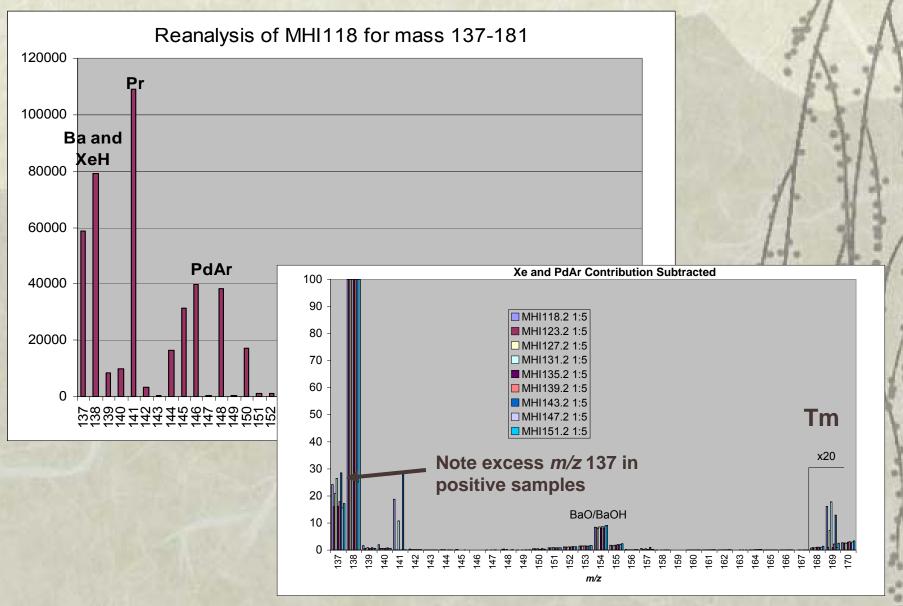
Results from Environmental Survey



Excess ¹³⁷Ba and Thulium Found in Samples Origin??



Excess ¹³⁷Ba and Thulium Found in Samples Origin??



Conclusions for Transmutation

- Molecular interferents will confuse isotopic patterns
 SIMS be wary
 - -ICP-MS better but not perfect
- Apply the Prime Directive
- Evidence for transmutation of Cs to Pr in <u>THIS</u> data set is weak

For NRL, Chart of Happiness for Transmutation is:

Off the chart!

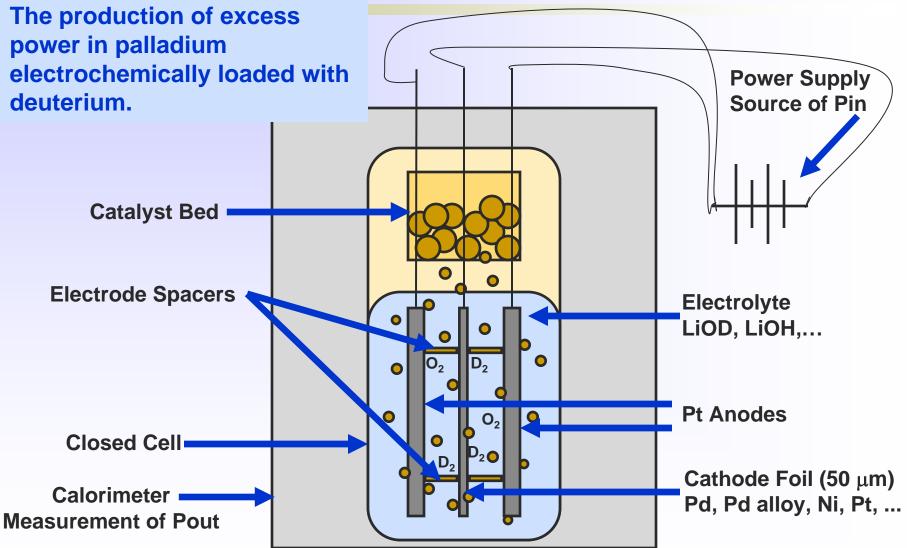
Heat



Short answer is Most defensible result

Electrochemical Experiments





Electrode sandwich configuration based on ENEA / Violante design



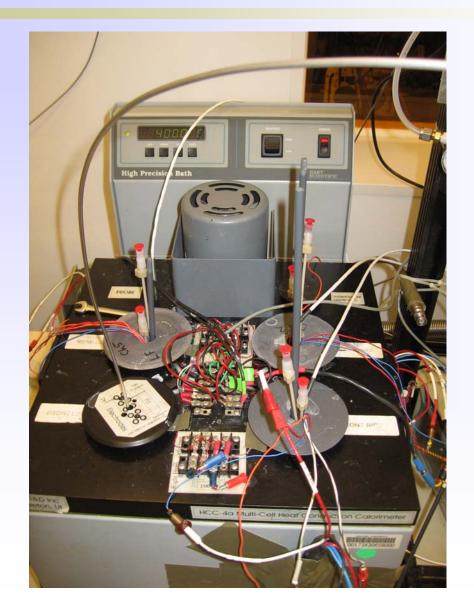




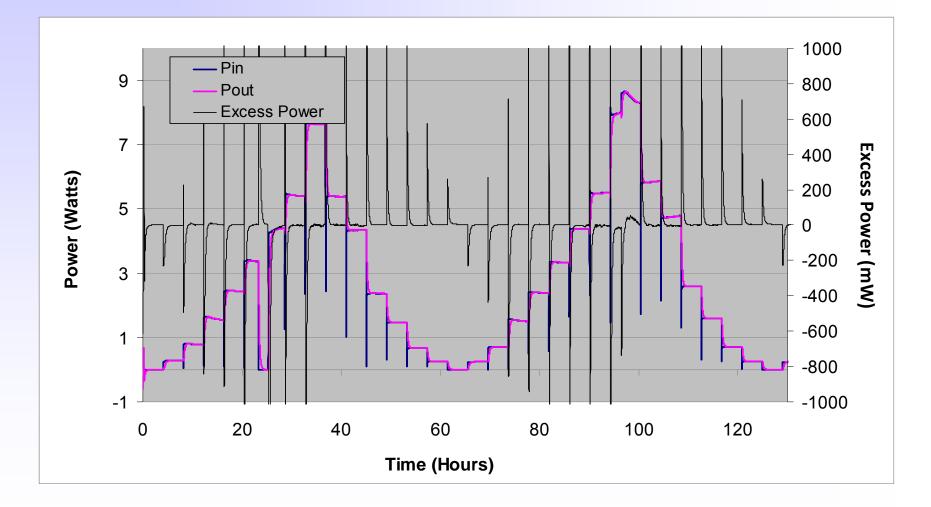




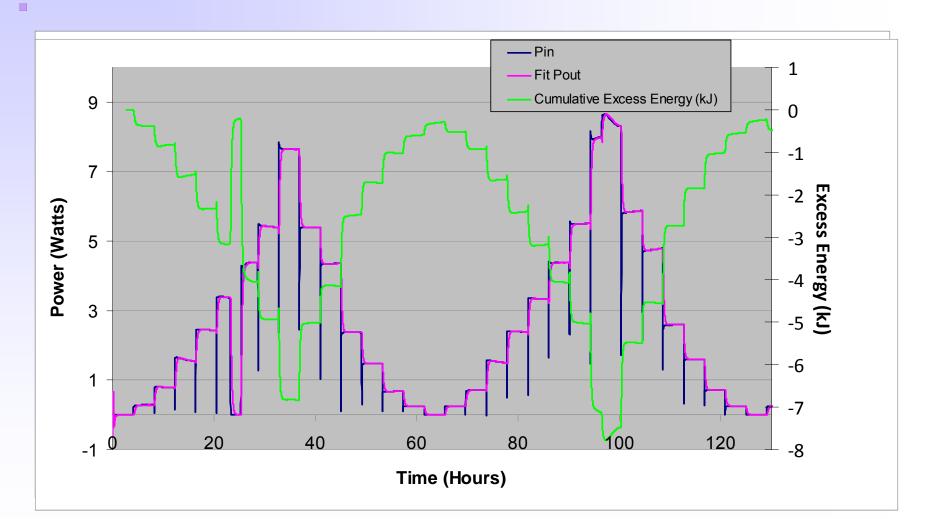




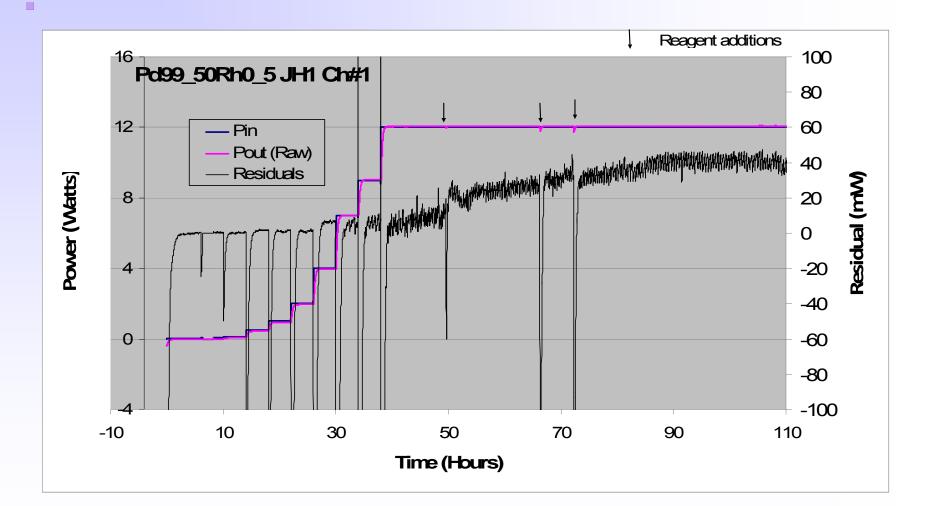
You recover the initial energy to heat cell Cathode #OL2 1-29-13 using LiOD and cycled



You recover the initial energy to heat cell Cathode #OL2 1-29-13 using LiOD and cycled



Unbalance after Additions Adding Raney Nickel often results in ~50 mW extra power

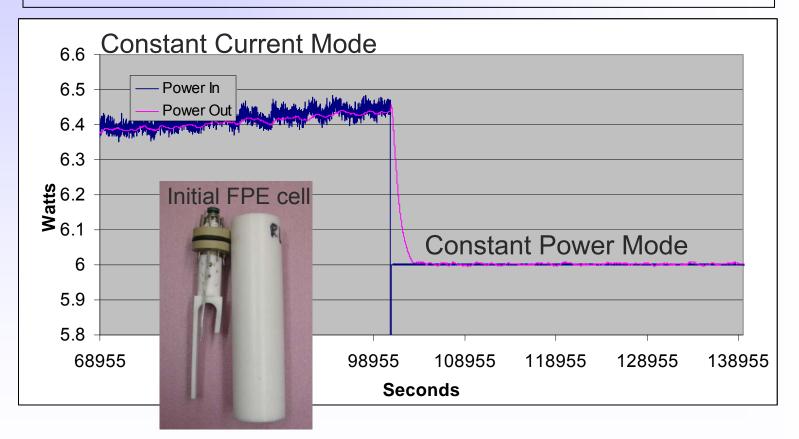




Electrolytic Loading Experiment

Original Fleischmann and Pons Approach

Many experiments, over 24 months, with consistent results
 > Power_{in}= Power_{out}



Teflon cell with Teflon internal parts and PEEK top (no capability for chemical additions)

Cathode Materials Investigated

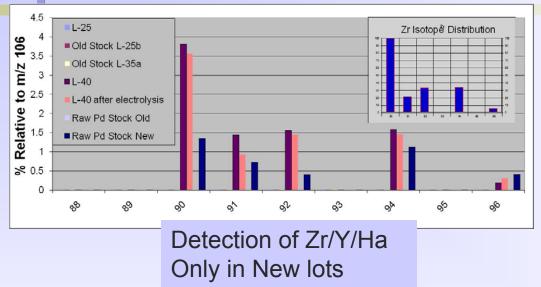


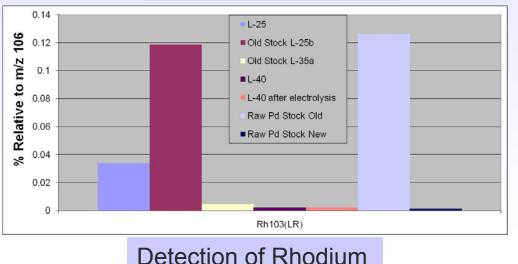
Cathoo	le Material	Number	x/LiOD	Pd 90%/ Rh10%		125
Pd/LiOD				1M LiOD, Pt anod	les, VSP	61
Goodfellow		22		new calorimeter		18
Platexis		19		BOPs		13
Engelhard		23		G&S Pd		12
						10
						8
160						3
		_				10
140 -						3
			📕 Jan 20	09-Mar 2011		2
100						2
Pd/LiOH 120 -						4
						2
100						2
						2
80 -						1
						1
60 -						1
						1
nisc 40						1
						1
						1
20 -						11
						4
0 -			I			3
	Negat	ives	Positives		4	
						2
				Ni		7
				Ni/Pd		2
				Ni/Pd/Ni		1
				Nb		3
				Та		2
				Total		<u>198</u>
				Grand Total		307

Why focus on Pd 90%/ Rh 10%?

Trace Impurities Inductively Coupled Mass Spectrometric Analysis







Only in Old lots

- Older lots of Palladium, that appeared to produce substantial heat, likely had only ONE source – Engelhard
- ICP-MS analysis shows different impurity profiles than current palladium lots
 - Older lots appear to have recycled Pd from catalytic converters
 - Current lots are much purer in these elements but have zirconium, yttrium, and hafnium present
 - Likely change in crucibles for melting to zirconia
 - Rhodium prices may drive recovery as a separate element
- Are the impurities responsible for the Fleischmann-Pons Effect?

Cathode #1 - Eve

14

12

10

8

6

4

2

0

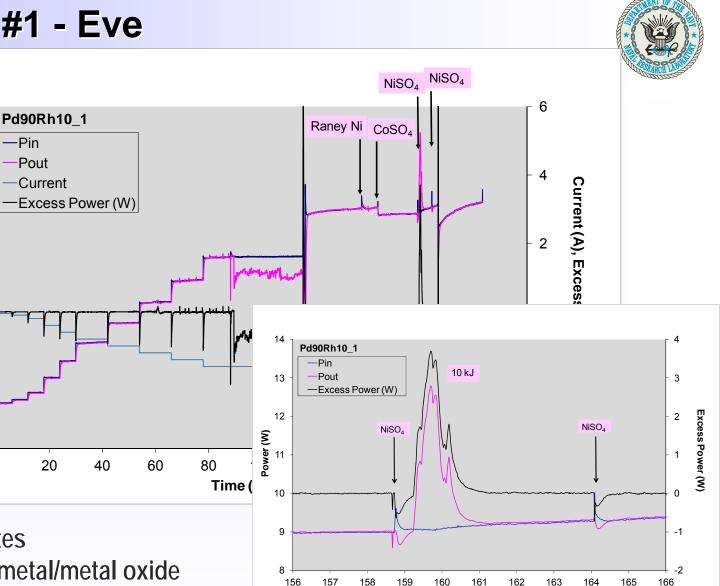
-2

0

Power (Watts)

Pin

Pout



Time (Hours)

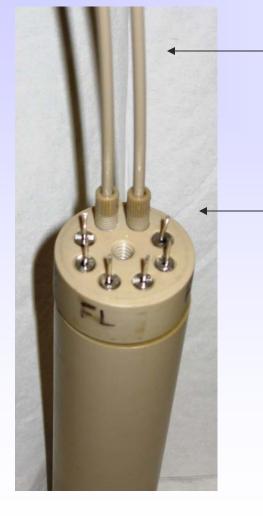
- 10 kJ over 90 minutes
- Event triggered by metal/metal oxide additions into 24 mL electrolyte volume??

20

• 5-10 mg NiSO₄ → 155 MJ/mole or 1.6 keV/molecule

All PEEK Cell Design





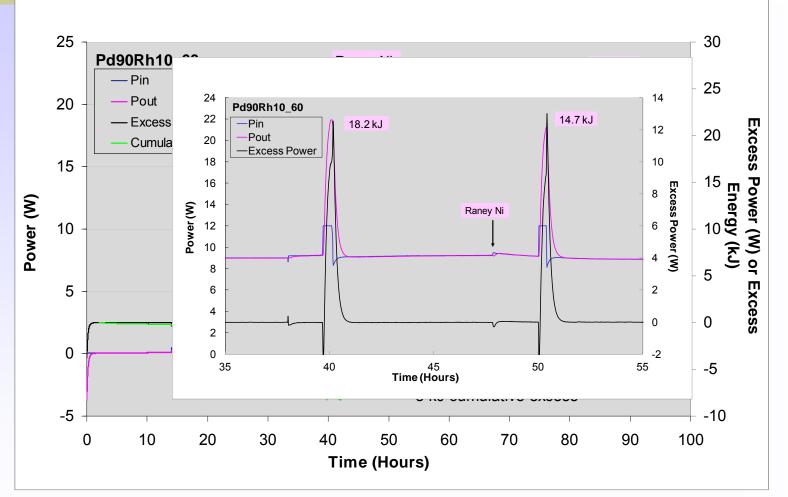
Tubes for chemical additions and pressure measurements

Cell top with Platinum feed throughs and High Pressure Liquid Chromatography fittings for sealing



Cathode #60





- Excess energy >> stored chemical energy
- Excess energy appeared before first addition
 - May have been due to cleanliness of cell

Production of large excess energy in a FPE is Rare!



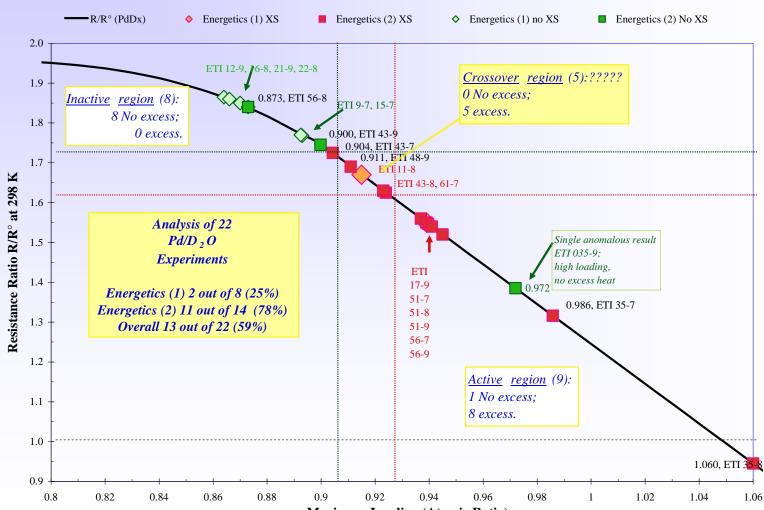
Ed Storms
 5/90
 SRI
 4/26
 NRL
 4/61
 Energetics
 ca. 8/800
 F&P

>1/?? described

SRI Results

Michael C.H. McKubre and Francis L. Tanzella, "New Physical Effects In Metal Deuterides" Contract No. HR0011-05-C-0089

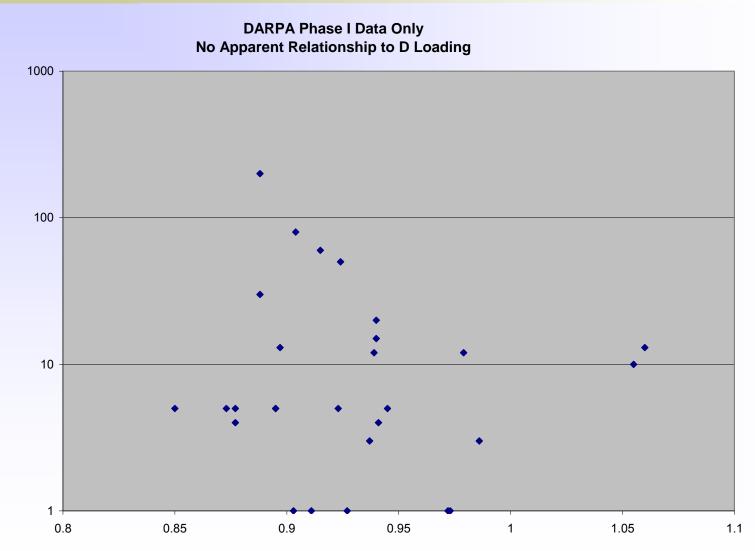




Maximum Loading (Atomic Ratio)

SRI Results No Apparent Relationship to D Loading





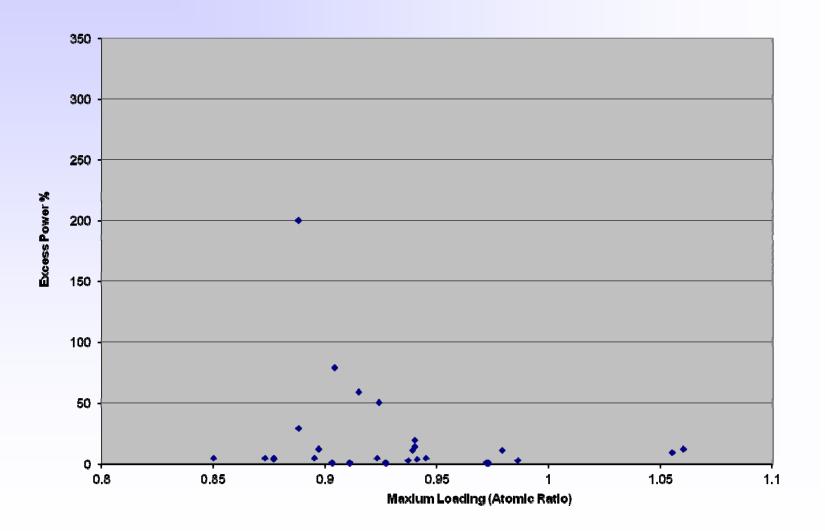
Maxium Loading (Atomic Ratio)

Excess Power %

SRI Results No Apparent Relationship to D Loading



DARPA Phase I Data Only



Instrumental Artifact? Shorts





Shorted in a certain way, RF generated

- Right frequency and similar pattern
- Very irreproducible not all shorts show RF
 - Need just the right cell resistance and mode in the VSP

> Shorted in a certain way, the VSP will:

- Misreport the power into the cell
- Power nearly sufficient to make observed effects
- Show no error messages





NRL's electrochemical shows excess energy that is:

Not chemistry

- But possibly an instrumental artifact
- For NRL, Chart of Happiness for heat is:

At 25%!

Gas Loading

D.A. Kidwell Naval Research Laboratory Washington, DC 20375 (202)767-3575 David.Kidwell@nrl.navy.mil

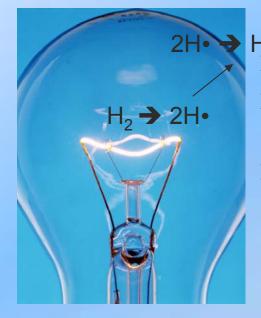
17th International Conference on Condensed Matter Nuclear Science – Daejeon, Korea



Langmuir



Worked for GE – wanted better light bulb



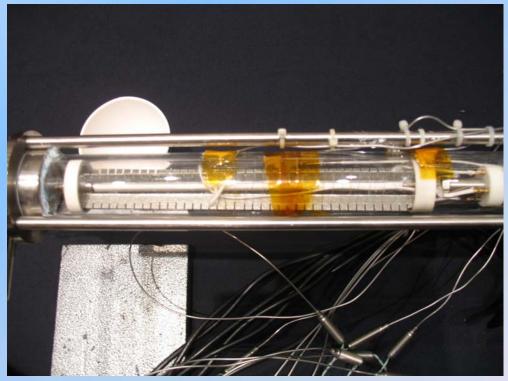
- H₂ Tested various gases
 - Observed extra energy coming from hydrogen
- *i.e.* surface of bulb much hotter with
 H₂ than any other gas

Eventual explanation:

- Appearance of extra energy but not really
 - H₂ splits into H• radicals on hot filament
 - Radical diffuse to cooler surface cannot recombine in gas phase
 - Radicals recombine at glass surface
 - Transfer lots of energy

Is this happening here?





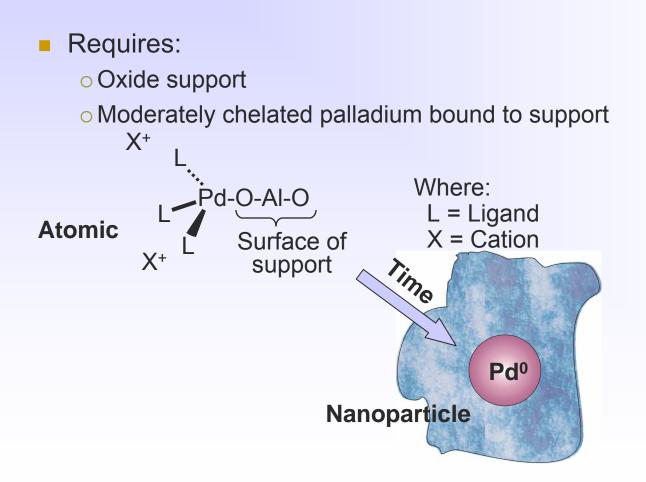
- Not a hot filament
- Is the "catalyst" just a better at making hydrogen radicals?
 - Ni and Pd known to form hydrogen radicals
- Accurate calorimetry will determine

I. Chorkendorff, J.N. Russell, Jr., and J.T. Yates, Jr., "Hydrogen Implantation in Ni(111) - A Study of H2 Desorption Dynamics from the Bulk", *Surface Science*, **182** (1987) 375-389.

Concept – Grow Particles in situ

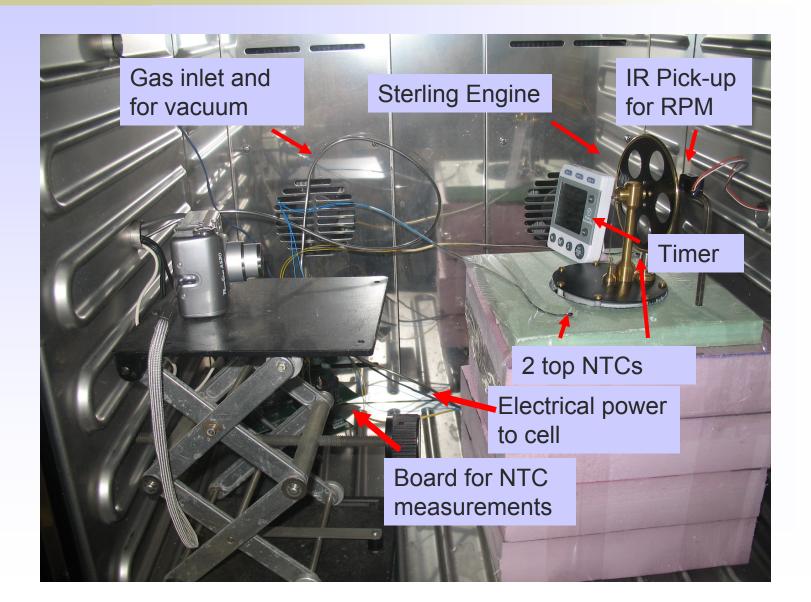


Use chemistry to modify particle growth
 Dispersed ions
 Nanoparticles



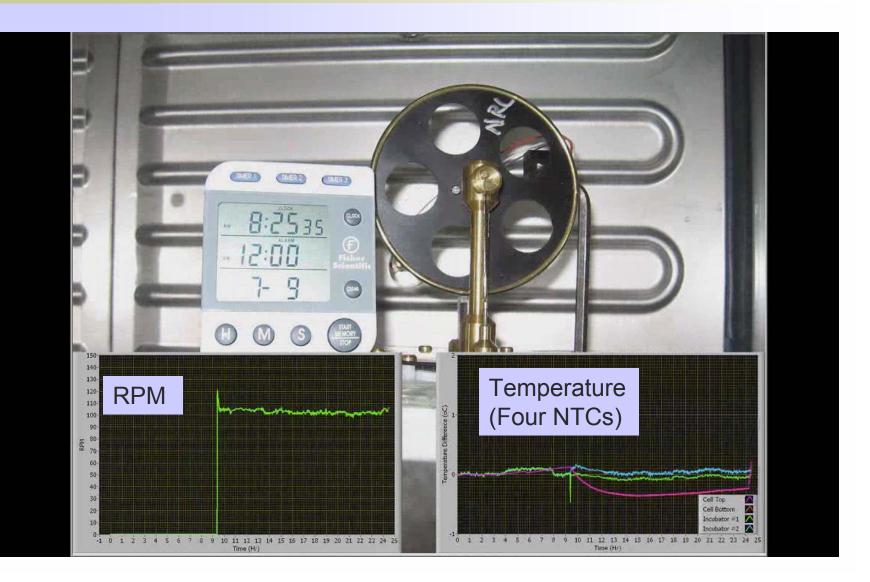
Useful Work? Movie showing D₂/Pd NP <u>partially</u> running a Sterling Engine

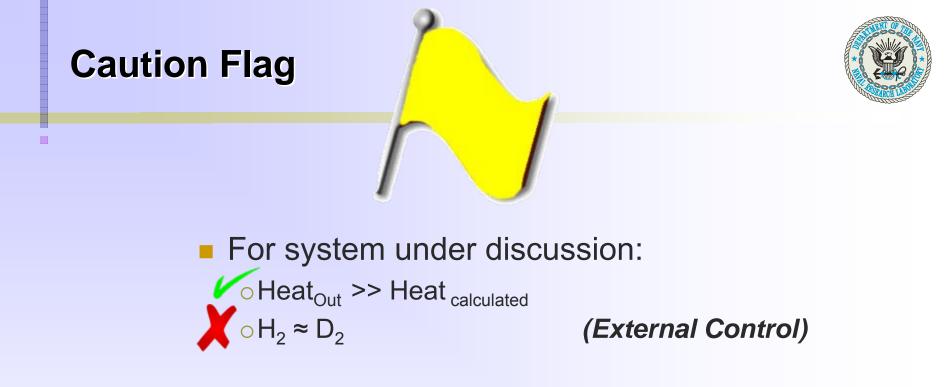




Useful Work? Movie showing D₂/Pd NP <u>partially</u> running a Sterling Engine







For NRL, Chart of Happiness for gas loading is:

At 50%

Summary - What Has NRL Observed?



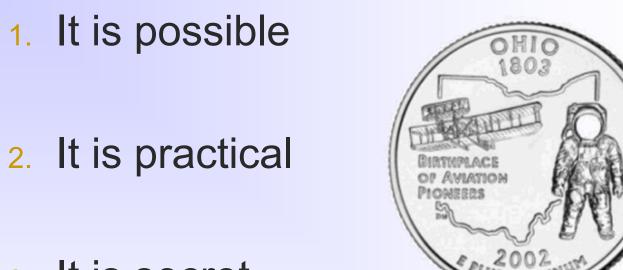
@NRL:

- No evidence of nuclear products
 - No X-Rays, neutrons, gammas, or radioactive materials
- No evidence for transmutation
- Evidence of excess heat in electrochemical cells
 Instrumental artifacts not ruled-out
- Evidence of excess heat in gas loading experiments
 All possible chemistry not ruled-out

Phenomenon deserves further study







3. It is secret

Wright brothers airplane - December 17, 1903

Curtiss - 1907 Curtiss Aeroplane Company - 1910 Curtiss-Wright Corporation -1929

Acknowledgements





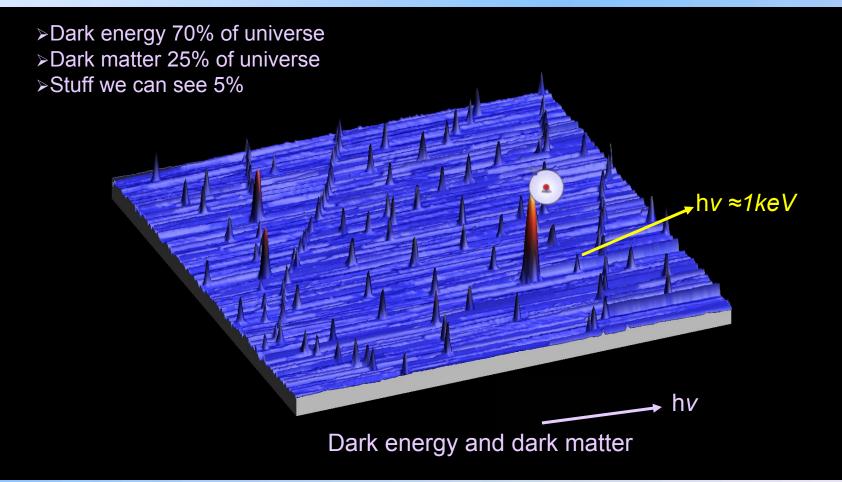
Off the Chart!

Questions

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Approved for Public Release, Distribution Unlimited NRL Approval #13-1231-2325



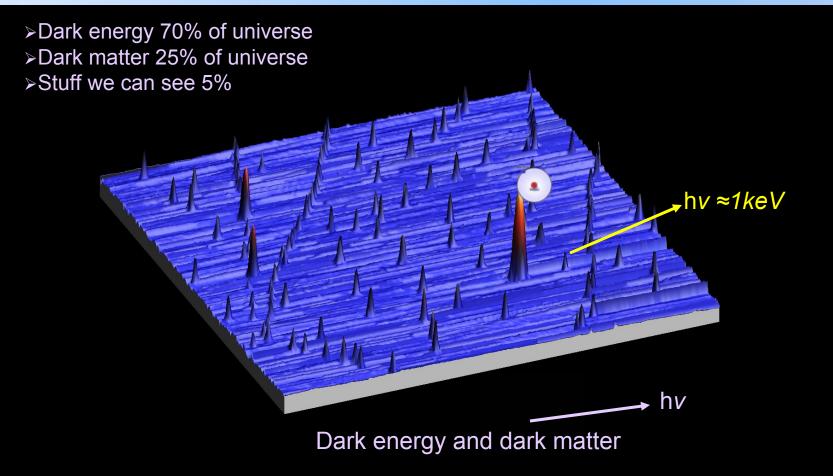


> Why 1 keV?

- Difficult energy to detect as non-penetrating and easily absorbed
- In electronic noise
- Film possible but issues with hydrogen radicals

Theory



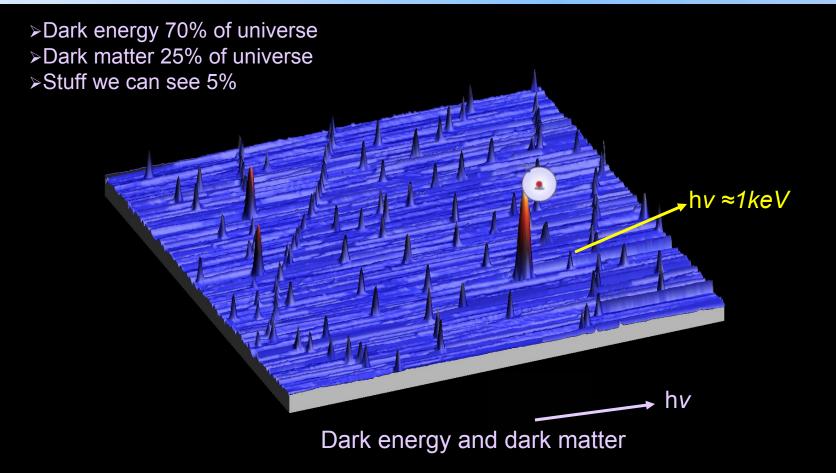


How many protons?

- 1g = 600W for 48 hr or 104 MJ
- Issue is this low weight loss would look like a leak

Theory Makes predictions but difficult experiments to confirm





- Violates principle of macroscopic reversibility
 - Normal protons can interact with catalyst to make dark matter
 - Dark matter cannot interact with normal matter to reverse process
 - Pumping X-rays onto catalyst could make protons at very low level
 Difficult to distinguish from normal water background