

SIXTEEN VALENCE-ELECTRON SPECIES CONTAINING BORON:
STRUCTURAL DIVERSITY ABOUND

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ABSTRACT

[B, C, F, H₂] and [B, N, O, H₂] have been investigated via high level *ab initio* calculations. Both cyclic and acyclic connectivities have been examined for the singlet state. Chemical intuition suggests a ketene-like structure for each system, with the general form H₂XBY, where X = C or N and Y = F or O. The calculations yield a variety of possible structures for these species, including both cyclic and the expected acyclic, hydrogen-bonded and fluorine-bridged. Eight minima have been located for [B, C, F, H₂] at both calculational levels. Fifteen minima have been located for [B, N, O, H₂]. Barriers to the rearrangements between isomers in the [B, C, F, H₂] system have also been evaluated. AIM and NBO analyses have been carried out to gain an understanding of the bonding in the various isomeric forms of [B, C, F, H₂] and [B, N, O, H₂].