



## — Nitrogen in the Environment — Leaching

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**W**hen soil becomes excessively wet through rainfall, the soil will reach a point where it can not hold any more water. This happens because the air spaces between soil particles become filled with water. As these air spaces fill, gravity will cause water to move down through the soil profile. (It can also pound or run off the surface of the soil.) As water moves down through the soil, nitrogen can be carried with it. This is called **leaching**.

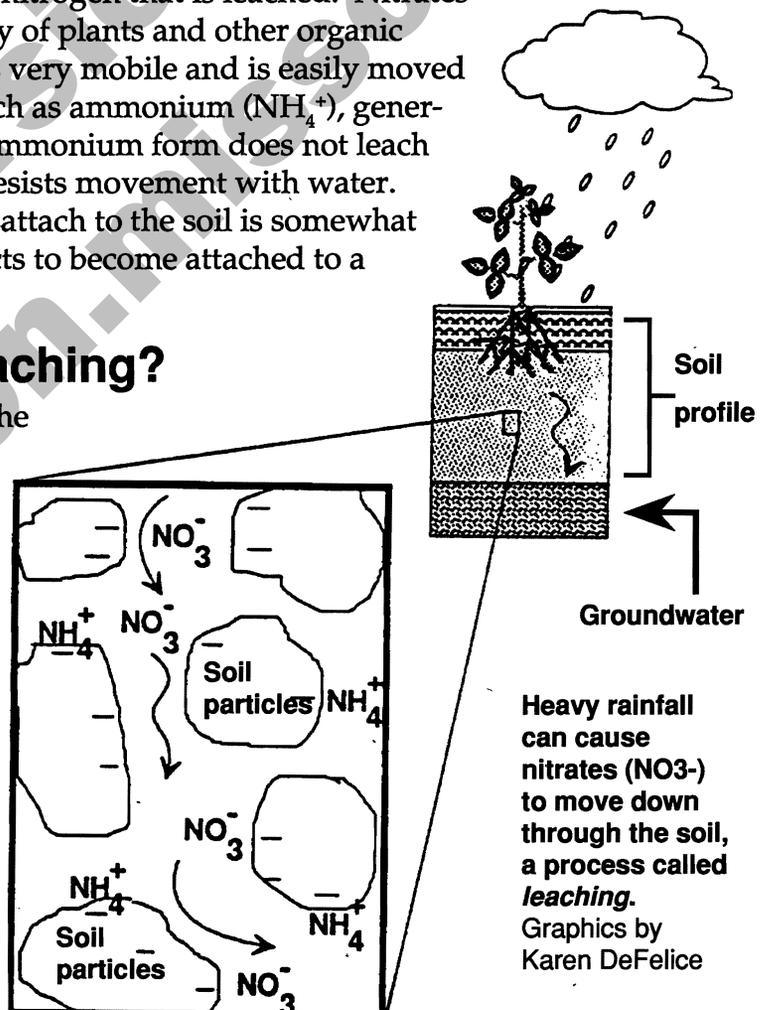
### Forms of Nitrogen that Leach

Nitrate ( $\text{NO}_3^-$ ) is primarily the form of nitrogen that is leached. Nitrates may originate from manures, the decay of plants and other organic materials or from fertilizers. Nitrate is very mobile and is easily moved by water. Other forms of nitrogen, such as ammonium ( $\text{NH}_4^+$ ), generally do not leach. Unlike nitrate, the ammonium form does not leach because it is attached to the soil, and resists movement with water. The forces which cause ammonium to attach to the soil is somewhat like the forces which cause metal objects to become attached to a magnet.

### What Affects Nitrate Leaching?

An important factor which can effect the degree of leaching is how much water a soil can hold. For example, by their nature sandy type soils can not hold as much water as clay type soils. This means that leaching of nitrates will take place much more easily in a sandy soil compared to a clay soil. Often times, leaching is not a significant factor in heavy clay soils.

Other factors which can affect nitrate leaching include amount of rainfall, amount of water use by plants and how much nitrate ( $\text{NO}_3^-$ ) is present in the soil system.



# Impact to Water Quality

Nitrate leaching can have a direct impact on water quality. Nitrate is very mobile and easily leaches with water. Heavy rains can cause nitrates to leach downward in the soil below the root zone. Whether nitrates continue to leach downward, and into groundwater, depends on underlying soil and/or bedrock conditions, as well as depth to groundwater. If depth to groundwater is shallow, and the underlying soil is sandy, the potential for nitrates to enter groundwater is relatively high. However, if depth to groundwater is deep and the underlying soil is heavy clay, nitrates will not likely enter groundwater. In some cases where dense hardpans are present, nitrate leaching will not progress beyond the depth of the hardpan.

Once nitrates get into the groundwater, the greatest concerns are for infants, less than one year old, and for young and pregnant animals. High levels of nitrates can be toxic to newborns causing *anoxia*, or internal suffocation. Seek alternative water sources if nitrate levels exceed the health standard of 10 ppm nitrate-N. Do *not* boil water to eliminate nitrates. It **increases** nitrate levels, rather than decreases them. The most common symptom of nitrate poisoning in babies is a bluish color to the skin, particularly around the baby's eyes and mouth. These systems of nitrate toxicity are commonly referred to as the "blue-baby" syndrome.

The potential of nitrates from animal manures and nitrogen fertilizers getting into groundwater can be reduced through good management practices. Applying manures and nitrogen fertilizers, when crops are actively growing, and using nitrates for growth and development, will reduce the amount of nitrate in the soil system, and thus the amount that could potentially be leached. However, little can be done to minimize the leaching of nitrates into groundwater that result from the ongoing decay of organic matter in the soil. In this situation, nitrates are simply a natural biological result of an ongoing cycle of nitrogen transformation processes in the soil.

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